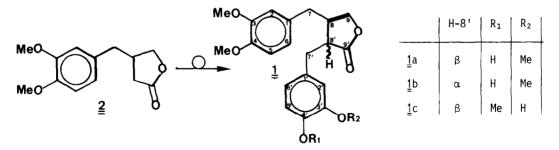
Prestegane A, from *Steganotaenia araliacea* Hochst.: The First Natural Dibenzylbutanolide Lignan with a *meta*-Phenol - A Short Synthesis of *cis* and *trans* (+) Arctigenin -

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<u>Summary</u>: A potential biogenetic precursor of steganes was isolated and its structure $\underline{1}$ c was determined using high resolution PMR, total synthesis and comparison with (±) arctigenin resulting from an unequivocal synthesis.

We have recently described six lignans from an antileukemic fraction of the West African variety *Steganotaenia araliacea* Hochst. (1) (2). A careful examination in TLC of the polar chromatographic fraction resulting from our previous work, revealed some closely related compounds. Preparative layer chromatography (Silicagel Merck Si 60, F 254; CHCl₃-Et₂0 1:1) vielded an amorphous compound which exhibited the following constants :

 $[\alpha]_{D}^{21} = -31^{\circ}$ (C 0,39, CHCl₃); IR (CHCl₃) 1770 (ν lactonic C=0), 3500 (ν phenolic OH) cm⁻¹ M⁺ calcd for C₂₁H₂₄O₆ 372.1573, found 372.1578; the PMR data (3) showed that it was a dibenzylbutanolide lignan identical or closely related to arctigenin <u>1</u>a (4)



In view of conflicting data concerning the aliphatic hydrogen assignments of the latter in PMR (5), we synthesized it using two independent short sequences from the known β -veratryl γ -butyrolactone $\frac{2}{2}$ as a common precursor. For the first route, $\frac{2}{2}$ was submitted to alkylidenation by 0-benzyl vanillin according to a Stobbe-like reaction (MeONa/C₆H₆, r.t., 120 mn ; 10 % H₂SO₄, r.t., 60 mn) followed by hydrogenation-debenzylation (Pd-C/AcOEt, 12 h) to give *cis* (±) arctigenin <u>1</u>b in a 43 % overall yield. This lactone <u>2</u> was quantitatively isomerized (KOH/MeOH, r.t.; 4 days, refluxing AcOH, 2 h) into the more stable lactone *trans* (±) arctigenin <u>1</u>.

Alternatively, a direct condensation of 0-benzyl vanillyl bromide (in THF-HMPT) on the anion of $\frac{2}{2}$ (LDA/THF, -78°, 2 h) followed by deprotection (Pd-C/AcOEt, 12 h) resulted in the same above *trans* lactone $\frac{1}{2}$ in a 68 % overall yield from $\frac{2}{2}$.

Still, our unknown lignan cleanly differed from both of these structures $\underline{1}a$ and $\underline{1}b$. Furthermore, alternate irradiations of every benzylic methylen protons at 400 MHz, gave a clear enhancement with sharpening of the corresponding pair of ortho-aromatic protons, which eliminated any substituents at C-2, C-6, C-2' and C-6'. Observation of the fragment at m/e 194 allowed to place phenol on the lignan (C_6-C_3) moiety attached to the carbonyl side as in arctigenin <u>1</u>a (6). At this stage, only lacton <u>1</u>c could be considered. Finally the structure of this new lignan was unequivocally proved with the help of the same direct synthetic previous sequence using 0-benzyl isovanillyl bromide instead of 0-benzyl vanillyl bromide (7).

According to these data, we named $\underline{1}c$ (-) prestegane A as a possible biogenetic precursor of lignans of the stegane group (8).

REFERENCES AND NOTES

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- Including the four lignans formerly discovered in a sample of other origin : S.M. KUPCHAN, R.W. BRITTON, M.F. ZIEGLER, C.J. GILMORE, R.J. RESTIVO and R.F. BRYAN, <u>J. Am. Chem. Soc.</u>, <u>95</u>, 1335 (1973). This paper is dedicated to memory of the late S. Morris KUPCHAN.
- 3. 250 MHz PMR (δ , CDCl₃) 6.78 (d, 1H, $J^{5'-6'} = 8.2 \text{ Hz}$) H-5'; 6.77 (d, 1H, $J^{5-6} = 8.2 \text{ Hz}$) H-5; 6.75 (d, 1H, $J^{2'-6'} = 2.1 \text{ Hz}$) H-2'; 6.63 (dd, 1H, $J^{6'-2'} = 2.1 \text{ Hz}$) H-6'; 6.58 (dd, 1H, $J^{6-5} = 8.2 \text{ Hz}$, $J^{6-2} = 2.1 \text{ Hz}$) H-6; 6.50 (d, 1H, $J^{2-6} = 2.1 \text{ Hz}$) H-2; 5.6 (broad s, 1H) phenol; 4.11 (dd, 1H, $J^{9b-9a} = 9 \text{ Hz}$, $J^{9b-8} = 8 \text{ Hz}$) H-9b; 3.91 (dd, 1H, $J^{9a-9b} = 9 \text{ Hz}$, $J^{9a-8} = 7 \text{ Hz}$) H-9a; 3.90, 3.84 and 3.83 (9H, 3s) 3 0Me; 3.00 (dd, 1H, $J^{7}b^{-8'} = 4.8 \text{ Hz}$, $J^{7}b^{-7'}a = 14 \text{ Hz}$) H-7b; 2.83 (dd, 1H, $J^{7'}a^{-7'}b = 14 \text{ Hz}$, $J^{7'}a^{-8'} = 6.2 \text{ Hz}$) H-7a; 2.61 (dd, 1H, $J^{7b-7a} = 13.5 \text{ Hz}$, $J^{7b-8} = 5.4 \text{ Hz}$) H-7b; 2.54 (m, 2H) H-8 and H-8'; 2.48 (dd, 1H, $J^{7a-7b} = 13.5 \text{ Hz}$, $J^{7a-8} = 7.8 \text{ Hz}$) H-7a.
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- 7. To give pure amorphous (±)-1c in a 70 % overall yield from 2. 90 MHz-PMR and IR (CHCl₃) spectra of natural and synthetic samples of 1c were strictly superimposable. Rf's in TLC using five solvant systems were identical.
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